

KEH, Z.

Decree of the Minister of Heavy Industry, no.129 of July 26, 1961 regarding a closer cooperation between sections of the Ministry of Heavy Industry and technical associations united in the Chief Technical Organization. Przegl mech 20 no.18:571 S '61.

Immunology

BULGARIA

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000721420009-6"

KEHAYOV, I., Laboratory of Cytoimmunology, Institute of Microbiology, Bulgarian Academy of Sciences

"Studies on Antigenic Relations Between Guinea Pig Kidney and Lung"

Sofia, Doklady Bolgarskoy Akademii Nauk, Vol 19, No 12, 1966, pp 1219-1222

Abstract: [English article] The treatment of guinea pigs with rabbit anti-guinea-pig-kidney serum raises the percentage of localizations of the tuberculous infection in the kidney (I. R. Kekhayov, "rvi kongres na mikrobiolozite v B"lgariya, Izd. BAN, 1965). The same phenomenon was established, although to a lesser degree, in the treatment of guinea pigs with heterologic anti-lung serum, which may be ascribed to the existence of antigen(s) common to kidney and lungs. The present communication contains some results of further studies on the antigenic relationship between the guinea pig's kidney and lung. Following a brief description of the materials (young rabbits) and methods used, it gives a detailed description and discussion of the results. An analysis of the data indicates that the complement-fixation test and the agar-gel precipitation test disclose that the antigenic mosaic of the guinea pig lung contains an antigen(s) common to the kidney antigens. This antigenic kinship of the two organs is considered significant for the specific organotropism of the tuberculous infection. References: 2 Bulgarian, 1 Soviet, and 7 Western. (Manuscript 1/1 received 7 Jul 66.)

KEHESY, A.

Color vision. Szemeszet 90 no.1:9-32 Feb 1953. (CLML 24:5)

1. Doctor Medical Sciences.

KEHIAIAN, Henry

✓ General study of the protolytic equilibrium. I. The chemical equilibrium in aqueous solutions of a monobasic and a monoacidic base. Henry V. Kehiaian. *Acta Acad. popolare Romane, Studii Cetn. Chim.* 3, 113-34 (1953).
 Math. Based on 6 definitions and 4 assumptions, a general equation is derived for aq. solns. of a monobasic acid and a monoacidic base, from which special and approximation formulas of the literature may be derived. *W.L.*

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RUMANIA/Physical Chemistry - Solutions, Theory of Acids and Bases

Abs Jour: Referat. Zhurnal Khimiya, No 2, 1958, 3935.

Author : Henry V. Kehiaian

Inst : APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000721420009-6"

Title : General Study of Protolytic Equilibrium. II. Chemical Equilibrium in Aqueous Solution of Acid and Base Mixture.

Orig Pub: Studii si cercetari chim., 1957, 5, No 1, 35-49.

Abstract: The equation of chemical equilibrium and its application to solutions of acids, bases and amphoteric substances (protolytes) was studied. The case of aqueous solutions produced by a mixture of polybasic acids and polybasic bases is discussed. The derived equation is a generalization of the equation derived earlier (part I, RZhKhim, 1956, 32081).

Card : 1/1

-7-

RUMANIA/Physical Chemistry - Solutions, Theory of Acids and Bases.

B-11

Abs Jour: Referat. Zhurnal Khimiya, No 2, 1958, 3936.

Author : Henry V. Kehiaian.

Inst :

Title : Composition of Boric Acids and Alkali Borates in Aqueous Solutions. I. Appendix to General Study of Protolytic

KEHIAIAN, H.

Studies on the liquid-liquid equilibrium. Pts. 1-3. Biul chim PAN
10 no.10:569-589 '62.

1. Institute of Physical Chemistry, Polish Academy of Sciences, Warsaw.
Presented by W. Swietoslawski.

KEHTIAN, H.

Thermodynamics of chemically reacting mixtures. Pts. 1-3.
Bul chim PAN 11 no.8:479-496 '63.

1. Institute of Physical Chemistry, Polish Academy of Sciences,
Warsaw. Presented by W. Swietoslawski.

KEHIAIAN, H.; SOSNOWSKA-KEHIAIAN, K.

Thermodynamics of chemically reacting mixtures. Pt.4. Bul
chim PAN 11 no.9:549-556 '63.

1. Institute of Physical Chemistry, Polish Academy of Sciences,
Warsaw. Presented by W. Swietoslawski.

KEHIAIAN, H.

Thermodynamics of chemically reacting mixtures. Pt. 7.
Bul chim PAN 12 no. 1:77-83 '64.

1. Institute of Physical Chemistry, Polish Academy of Sciences,
Warsaw. Presented by W. Swietoslawski.

KEMTAJAN, H.; FAJANS, A.

Thermodynamics of chemically reacting mixtures. Pt. 8. Bul chim PAN 12 no.4:255-262 '64.

1. Institute of Physical Chemistry, Polish Academy of Sciences, Warsaw. Presented by W. Swietoslawski.

GRIGORIAN, R.

Thermodynamics of chemically reacting mixtures, pt. 9. Bull chim PAN 12 no. 5:323-329 (1964).

P. Institute of Physical Chem., Acad. Acad. of Sciences, Warsaw. Presented by Z. Swietoslawski.

KERIATAN, H., SŁOMKOWSKA-KERIATAN, K.

Thermodynamics of crystallinity resulting in values - Pt. 10-11.
Bull. Chem. PAN 12, no. 6, 425-439, 1964.

1. Institute of Physical Chemistry of the Polish Academy of
Sciences, Warsaw. Submitted April 17, 1963.

KEHIAIAN, H.

Thermodynamics of chemically reacting mixtures. Pt.12. Bul
chim PAN 12 no.7:497-501 '64.

1. Institute of Physical Chemistry of the Polish Academy of
Sciences, Warsaw. Submitted May 27, 1964.

KEHIAIAN, H.

Thermodynamics of chemically reacting mixtures. Pt.13. Bul
chim PAN 12 no.8:567-573 '64.

1. Institute of Physical Chemistry of the Polish Academy of
Sciences, Warsaw. Submitted June 20, 1964.

KEHIAIAN, H.

Thermodynamics of chemically reacting mixtures. Pt.14. Bul
chim PAN 9[i.e. 12] no.9:675-679 '64.

1. Institute of Physical Chemistry of the Polish Academy of
Sciences, Warsaw. Submitted July 16, 1964.

KEHIAIAN, K.; SOSNOWSKA-KEHIAIAN, K.

Thermodynamics of chemically reacting mixtures. Pis. 5-6.
Bul chim PAN 11 no.10:583-596 '63.

1. Institute of Physical Chemistry, Polish Academy of Sciences,
Warsaw. Presented by W. Swietoslawski.

Kehl, Jerry.

Distr: 4E2c(j)/4E3d

✓ Formaldehyde condensation. I. Stanislaw Malinowski, Jerzy Kehl, and Stanislaw Tyrlak (Politechnika, Warsaw), Roczniki Chem. 34, 591-400 (1960) (English summary).—Condensation of HCHO (I) to polyalc. aldehydes was studied in presence of various catalysts. PbO (II), obtained by decomprn. of Pb orotate at low temps., was found to form an active complex with I, whereas II prep'd. at high temps. was inactive. Oxides of other common metals (except CaO and MgO) were inactive as well as organolead compds. contg. the Pb(OH) group, or complexes of II with glycol or glyceraldehyde, acetylacetone, or acetacetates.

A. Kreglewski

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1-BW(BW)
1-FAJ(NB)
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be

Zaklad Technologii Organicznej I Politechniki, Warszawa.

KEHLER, Miro

Development of telegrams from the beginning of society up to our days.
PTT zbor 16 no.1/2:34-36 F '62.

KARSAY, Gyula, Dr.; KEHLI, Istvan, Dr.; KORANYI, Andras, Dr.

Pathography and therapy of intracerebral vascular attacks with
special regard to ACTH therapy. Orvn. hetil. 99 no.31:1049-1053
3 Aug 58.

1. A Janos Korhaz-rendelointezet (igazgato: Tako Jozsef dr.) I. sz.
Beloasztylyanak (foorvos: Koranyi Andras dr.) koslemenye.
(CEREBRAL HEMORRHAGE, ther.
ACTH intravenous drop infusion (Hun))
(ACTH, ther. use
cerebral hemorrh., intravenous drop infusion (Hun))

CA KEHREN, M.

27

The spontaneous combustion of oiled textiles. M. Kehren. *Textil-Rundschau* 3, 409-17 (1949); *Chem. Zentralbl.* (Russian Zone Ed.) 1949, I, 943.—The spontaneous combustion of oiled textiles is caused by the use of several unstd. compds., such as linoleic or linolenic acid, or their glycerides, in the oiling preps. Such preps. are best tested with the Mackey test, which gives the relation between spontaneous temp. increase and time. Six curves are given showing the behavior of textile olein, mineral oil, water-contg. fat-powd. dressing oils, and the influence of mineral soaps on such oils. Mineral oils are safe, even in the presence of metal soaps (Fe, Cr). The spontaneous oxidation can be retarded by anticalysts such as α -naphthol or hydroquinone. Further suggestions are offered for the prevention of spontaneous combustion. M. G. Moore

CA

KEHREN M.

29

Rubber poisons: origin and removal from textiles. M.
Kehren, *SVF Fachorgan Textilberedl.* (Basel) 4, 111-
1163-70 (1949).—The chemistry of natural and synthetic
rubbers, and chem. agents, e.g., Cu, Mn, Fe, Co, oils, and
acids, which promote their "perishing" on textiles are dis-
cussed. The sources and detection of rubber poisons, es-
pecially traces of Cu and Mn, are reviewed, and the use of
Trilon B for removing these metals is discussed. B. A.

KENDRICK, A. H.

Journal of the Science
of Food and Agriculture
Jan. 1954
Sanitation

Iron treatment of textile waste-waters. L. M. Kehren and H. Denks (Z. ges. Text. Ind., 1953, 55, No. 8, 417-426; T. Text. Ind., 1953, 44, 1551).—The removal of suspended matter from textile waste-water by means of treatment with Fe salts or metallic Fe (-waste) is still the subject of investigation, the nature of the colloid reaction involved being, as yet, uncertain. Possible physical and chemical processes involved are discussed and the application of the Fe treatment method to practical water purification, particularly in relation to textile processing effluent, is described.

R. B. CLARKE

KEHYYERDAL, Tur[Heyerdahl, Thor] (Norvegiya)

Seaways to Polynesia. Priroda 52 no.1:75-84 '63.
(MIRA 16:1)

(Polynesia—Discovery and exploration)

KEIBS, L.

~~SEARCHED, INDEXED~~

CONFIDENTIAL
35

PHASE I BOOK EXPLOITATION FOL/5981

Symposium on Electroacoustic Transducers. Krynica, 1958

Proceedings of the Symposium on Electroacoustic Transducers [held in] Krynica, 17-26 September, 1958. Warsaw, Państwowe Wydawnictwo Naukowe, 1961. 442 p. Errata slip inserted. 630 copies printed.

Sponsoring Agency: Polish Academy of Sciences. Institute of Basic Technical Problems.

Ed. in Chief: Janusz Kacprowski, Doctor of Sciences; Editing Committee: Ignacy Malecki, Professor, Doctor of Sciences; Wincenty Pajewski, Doctor; and Jerzy Wehr, Master of Sciences; Secretary: Julianz Mierzejewski.

PURPOSE: This book is intended for physicists and acoustical engineers.

COVERAGE: The book is a collection of detailed research papers constituting the proceedings of a conference held in Krynica from 17 to 26 September 1958 under the auspices of the Institute of Technical Problems, Polish Academy of Sciences.

Card 1/8

35

Symposium on Electroacoustic Transducers

FOL/5981

The following basic problems are treated: 1) theoretical research on energy transformation processes; 2) experimental development of new types of transducers; 3) electroacoustic measurements; 4) technology of piezoelectric and magnetostrictive materials; 5) construction of transducers for technical needs; and 6) design of acoustical transducer systems. No personalities are mentioned. References (if any) follow the individual articles.

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1. Classification of electromechanical transformation methods in the light of the tasks faced within [sic] the design and construction of electroacoustic equipment. V. S. Grigor'yev	7

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Symposium on Electroacoustic Transducers

POL/5981

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Card 6/8

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000721420009-6"

KEIETI, J.

Study of the biological properties of some species of bacteria
of the family Enterobacteriaceae. Cesk. farm. 13 no.3:110-114
Mr'64.

1. Katedra biochemie a mikrobiologie farmaceutickej fakulty
UK, Bratislava.

The creep performance of a few soft solders. A. Keil.
Metall II, 730-2(1957).—Nearly all Bi alloys and soft
solders show a greater hardness in the cast condition on both
short time and long time hardness tests. In all cases the
hardness falls rapidly with time of loading. This creep
behavior in the case of pure Sn can be improved by the
alloying of Ag and Sb. This is also true for Ag added to
Cd-Zn alloys. The use of such materials for soldering,
therefore, has advantages in technology. The advantage of
Ag added to Pb-Sn solders is not demonstrable; this element
is added, therefore, only if other reasons exist for its presence
in the alloy.

H. Stoerz

7
FEZC-1

MT

The synthesis of an isomer of dioscoraine. F. Šorm and B. Keil (Tech. Univ., Prague). *Collection Czechoslov. Chem. Commun.*, 12, 655-60 (1947) (in English).—The prep. of a dioscoraine (I) was attempted for structural studies. *Di-Me α,γ-di-2-pyridylglutarate* (III), prep'd. from Me 2-pyridinecarboxylate (II) by 2 methods, was isolated as the *diesterate*, m. 191-5° (from dioxane). (a) III (11 g.) in 20 cc. CaH_2 was added dropwise to 2.96 g. atomized K in 100 cc. CaH_2 and after 18 hrs. 9.52 g. CH_3I was added with spontaneous reaction; when the alk. reaction had disappeared after 3 hrs. at 80°, the soln. was shaken with 25 cc. 3 N HCl at 0°, the aq. ext. made alk. with satd. aq. KOH , and the product which sepd. ext'd. with Et_2O , giving 3.8 g. (34%) II, b. 214-20°, as well as 1.4 g. of a fraction b. 270°. (b) Dry $(\text{CH}_3)_2\text{O}$, (0.66 g.), 50 mg. piperidine, and 0.6 g. III were heated to 120° 5 min.; distn. gave 1.65 g. III, b. 120°, and 4.4 g. (84.5%) II, b. 231-10°. On hydrogenation 7.8 g. II in 100 cc. HOAc and 17.8 g. PtO_2 (Adams) at room temp. and 140 mm. took up 3335 cc. H (323) theory and gave 0.8 g. of a fraction b. 98-100°, and 3.4 g. b. m. 175°. The latter in 10 cc. CaH_2 was analyzed chromatographically on 75 g. neutral Al_2O_3 .

The following fractions were obtained (no. of fraction, cc. of developer, and mg. of residue given): 1, 15 cc. C_6H_6 , 675; 2, 15 cc. C_6H_6 , 430; 3, 15 cc. C_6H_6 , 280; 4, 30 cc. C_6H_6 , 230; 5, 30 cc. C_6H_6 , 140; 6, 30 cc. C_6H_6 , 225; 7, 40 cc. C_6H_6 , 115; 8, 40 cc. C_6H_6 , 80; 9, 40 cc. C_6H_6 , 75; 10, 40 cc. C_6H_6 , 75; 11, 40 cc. C_6H_6 , 50; 12, 40 cc. CHCl_3 , 275; 13, 80 cc. CHCl_3 , 600; 14, 40 cc. CHCl_3 , 25. Fractions 2-6 were colorless gums, cryst. on wetting with H_2O . Fractions 7-11 were fluorescent noncryst. gums. Fractions 12-14 were dark gums, cryst. on standing after wetting with C_6H_6 . Fraction 1, a dark-colored gum, was reanalyzed on 18 g. Al_2O_3 ; the C_6H_6 eluate gave a dark gum with a basic odor. The CHCl_3 eluate was combined with fractions 2-6; the crystals produced, recrystd. from Et_2O , sublimed at 0.001 mm., and recrystd. twice from Et_2O , m. 17°, analyzed as I. The mother liquor from I, on Al_2O_3 , gave a product m. 136-7° from the C_6H_6 eluate and fractions 13-14 gave a compd. m. 160-1° (from Et_2O), both of unknown structure. John W. Green

John W. Green

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C

The synthesis of sparteine and isosparteine. P. Šorm and B. Keil, *Czechoslovak. Chem. Commun.*, 13, 544-56 (1948) (in English).—The by-product in the prepn. of di-Me α,γ -di-2-pyridylglutamate (I) (C.A. 42, 6026) was shown to be Me 4-oxo-3-(2-pyridyl)-1-pyridocarboxylate (II) formed from I during distn. under reduced pressure by the simultaneous splitting off of MeOH and dehydrogenation. II, b.p. 270-8°, m. 100° (from C_6H_6 -petr. ether) [urate, m. 239° (decomp.) (from dioxane)], was obtained from Me 2-pyridineacetate (III) in 12% yield with CH_3O_2 , III (0.42 g.), 0.380 g. $\text{Hg}(\text{OEt})_2$, and 0.5 cc. ArO_2 were heated 2 hrs. at 125-130°, the HgOAc and excess ArO_2 distd., and the residue diatd. to give 2 fractions, b.p. 100° and b.p. 220° (IV), which crystall. IV in C_6H_6 was chromatographed on 3 g. neutral Al_2O_3 ; the first 100 cc. of C_6H_6 eluate gave yellow needles of II, m. 100°. The identity of the II obtained by the 2 methods was shown by mixed m.p. and identical absorption curves in the visible spectrum. I (7.8 g.) in 100 cc. glacial AcOH and 0.78 g. Adams PtO_2 at room temp. and 140 mm. Hg excess pressure of II absorbed 3.235 cc. II and was worked up to give 0.8 g., b.p. 88-100°, and 3.4 g., b.p. 175° (V). V (7 g.) in thiophene-free C_6H_6 was passed through 210 g. Al_2O_3 . The following fractions of 200 cc. were collected and evapd. separately (fraction no., eluent, wt. in g.): 1, C_6H_6 , 0.32; 2-12, C_6H_6 , 3.05; 13-20, $\text{C}_6\text{H}_6\text{-CHCl}_3$, 0.08; 21-5, $\text{C}_6\text{H}_6\text{-CHCl}_3$, 0.87; 20-7, Et_2O , 0.20; 28-31, $\text{Et}_2\text{O-CH}_3\text{OH}$ (9:1), 1.9. Fraction

I did not crystall. Fractions 2-12 crystall. on wetting with Et_2O and on fractional crystn. from Et_2O gave 0.39 g. colorless prisms, m. 171° (VI), and 0.4 g. longish prisms, m. 131° (VII) (mixed m.p. showed a depression of 13°), while the monocryst. mother liquor on further chromatography gave 0.53 g. VIII. VI, m. 172° after 2 recrystn. from Et_2O and sublimation *in vacuo*, was identical with the dihydrosparteine previously described; VII, m. 133° after 2 recrystn. from Et_2O and sublimation *in vacuo*, was shown to be another dihydrosparteine isomer. Fractions 21-3 and 26-7 did not crystall. on wetting with Et_2O . Fractions 20-7 and 28-31 crystall. on wetting with C_6H_6 and were recrystn. from $\text{Et}_2\text{O-CH}_3\text{OH}$ to give Me 4-oxo-3-(2-pyridyl)octahydro-1-pyridocarboxylate (VIII). II (3.6 g.) in 70 cc. glacial AcOH and 0.4 g. Adams PtO_2 gave as above 3.4 g. of a colorless resinlike substance which was heated 2 hrs. at 140-210° *in vacuo*. The residue, 2.9 g., was monocryst. and after soln. in 10 cc. C_6H_6 was chromatographed through 120 g. neutral Al_2O_3 . Fractions of 100 cc. were taken and evapd. separately (fraction no., eluent, wt. in g.): 1, C_6H_6 , 0; 2-10, C_6H_6 , 1.31; 11-12, C_6H_6 , 0; 13-15, CHCl_3 , 0; 16-20, Et_2O , 0, and 21-4, Et_2O , 1.49. The cryst. substance, m. 118-25°, from fractions 2-12 was septd. into VI and VII; fractions 21-4 yielded VIII. VI (0.36 g.) in 10 cc. 30% H_2SO_4 was reduced electrolytically at 25-2° during 3.5 hrs. at 1.3 amp. with activated electrodes of pure Pb, made alk. with 1.5 g. solid $\text{Ba}(\text{OH})_2$, steam-distd., and the steam distd.

(cont.)

late made neutral to methyl orange and taken to dryness to give 0.24 g. HCl salt (IX). IX (0.24 g.) in 2 cc. hot H_2O was added dropwise to 0.51 g. Na picrate in 10 cc. H_2O , and the crude picrate which sepd. was washed with alc. and recrystd. from alc.-dioxane to give 0.44 g. of the picrate, m. 222° (decomp.), presumably the dipicrate of



sparteine (X). VII on similar treatment gave 2 picrates, m. 187-9° (from alc.-dioxane) (XI) and m. 215° (from alc.-dioxane) (XII), which were sepd. mechanically. XI may be the dipicrate of a isosparteine while XII is believed to be similar to X (cf. C.A. 28, 4064; 35, 3031). By examination of the structure of sparteine, if rings B and C are cis to each other, rings A and D can be attached to ring B or C, respectively, either cis or trans. Thus, 3 isomers are possible: (1) A: B cis, C: D cis; (2) A: B cis, C: D trans and (3) A: B trans, C: D trans. Each of these isomers will be resolvable into optical antipodes.

H. L. Yale

✓ A

4

Electrophoresis of proteins in agar jelly. A. H. Gordon, B. Kral, and K. Sebesta (Tech. Univ., Prague, Czech.). *Nature* 164: 488-9 (1949).—Perritin and hemoglobin were sepd. by electrophoresis in 1% agar jelly in 19 hrs. at 3.5 v./cm. Diffr. 1:10,000 pepsin at pH. 3 in NaOAc buffer sepd. into 2 main bands which was revealed by spraying with Pauli reagent (I) (Conden, *et al.*, *C.I.* 40,41009). The adsorption at 290 m μ corresponded to the color developed with I. Sepsis. of the proteins of egg white at pH 8.8 and 8.0 were similar to those obtained by the Tiselius technique. Earl S. McCollum

CA

AC

Electrophoresis of proteins in agar jelly. A. H. Gordon, J. Kell, K. Scherba, O. Kneval, and P. Burek (Tech Univ., Prague). *Collection Czechoslov. Chem. Commun.* **13**, 1-10 (1958) (in English), cf. *C.A.* **44**, 18342. An app. and methods for the electrophoretic sepn. of proteins in agar jelly (1) are described in detail (cf. *C.A.* **40**, 4102). The sepd. protein bands are detected either by their ultraviolet absorption at 280 or 290 m μ or by "printing off" the I on the filter paper and detecting with Pauli reagent or ninhydrin. Protein mobility at different agar concns. depends on the mol. wt. Whereas hemoglobin moves with the same speed, ferritin is slowed down considerably by higher I concns. Electrophoresis of egg white and normal human plasma resulted in sepn. of compounds similar to the sepn. obtained with the Thelin method. Crude pepsin was sepd. into different fractions. A drawback of the method is the difficulty of removing the last traces of I from amino acids and proteins. The chief use for the method will be the sepn. of proteins on a preparative scale. K. Scherba

CA

117

1 Protein and amino acids. IV. Partition chromatography of dinitrophenyl amino acids on kieselguhr and silicated materials. O. Kneval, B. Keil, A. Malý, and P. Šorm (Tech. Univ., Prague). *Collective Czechoslov. Chem. Commun.*, 15, 618-20 (1951) (in English); cf. *C.A.*, 45, 10182c, 46, 3587d. The substitution of kieselguhr for silica gel in the method of Sanger (*C.A.*, 40, 6551c) gives a more reproducible system. If the kieselguhr is silicone coated or a silicone polymer is used, a "reversed-phase" chromatography ensues. The rate of movement, R_f , as a function of pH was evaluated as the ratio of R for a reference substance to R for the dinitrophenyl amino acid. The pH is so selected that optimum sepn. is obtained for any given mixt. The dinitrophenyl derivs. of threonine, glycine, alanine, and phenylalanine and dinitrauniline were used as examples with a mobile phase of $\text{CHCl}_3\text{-EtOAc}$.

H. P. Block

CA

Proteins and amino acids. V. The preparation of apoferritin by alcohol precipitation. B. Keil and J. Pották-hová (Central Chem. Research Inst., Prague). *Collection Czechoslov. Chem. Commun.* 16, 204-6 (1951) (in English); cf. C.A. 46, 5022f.—Prepn. of ferritin (I) by Laufberger's procedure (C.A. 33, 20631) gave within 8-9 hrs. cryst. material which yielded pure I after a single further cryst. Cryst. I was dissolved in an acetate buffer (pH 4.85) and freed from Cd by dialysis against the same buffer. The resulting clear dark-brown soln. contained 0.67 mg. N and 1.44 mg. Fe/ml. This soln. (21 ml.) was treated with 200 mg. Na₂SO₄ (II) and 100 mg. phenanthroline (III), let stand at 1° overnight, a further 100 mg. II and 200 mg. III added, the soln. let stand 1 hr., cooled to -5° while 20 ml. EtOH was added gradually from a separatory funnel with the stem in the capillary and dipping below the liquid surface; the soln. was centrifuged 4 min. at -5° at 90, 000 r.p.m.; the liquid decanted, the sediment dissolved in the acetate buffer, made up to 14 ml., and 200 mg. II and 100 mg. III added; after 1 hr. a 2nd ptn. was carried out at -5° by addn. of 13 ml. EtOH; the centrifuged sediment dissolved in

buffer soln., and the apoferritin (IV) soln. dialyzed. faintly pink soln. of IV (50 mg.) contained 0.66% Fe and 18.9% N. The protein crystd. spontaneously on addn. of 1/2 vol. 30% CdSO₄ soln. to the dialyzed soln. VIII. A new synthesis of glutathione. J. Radlinger and F. Sorm. *J. Am. Chem. Soc.* 71, 214-19; cf. C.A. 46, 3867d.—Into S-benzyl-L-cysteine (I) (14 g.) suspended in 360 cc. abs. dioxane in a 1-l. flask fitted with a liquid-sealed stirrer, reflux condenser with guard tube, and a gas-delivery tube to the bottom, and heated to 40°. COCl was introduced at 40-5° until the I dissolved, then 3 hrs. in excess COCl was removed with dry air, the distill. off at 20-40° and 15 mm. after filtering, and the residue taken up with boiling dry C₆H₆; on cooling there was obtained 10.8 g. S-benzyl-N-carboxy-L-cysteine anhydride (II), m. 105-6°, $[\alpha]_D^{25} -43 \pm 1.5^\circ$ (c 1.34, dioxane); 77.5%, m. 105-6°, $[\alpha]_D^{25} -43 \pm 1.5^\circ$ (c 1.34, dioxane). To 2.75 g. II in 30 cc. tetrahydrofuran (III), cooled to -75° was added dropwise a precooled soln. of 1.36 g. H₂NCH₂CO₂H and 0.34 g. 1-methylglutathione in

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Proteins. Bohumil Kell, Chem. Zvesti 6, 483-7 (1953).—
A lecture. Jan Micka

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CIA-RDP86-00513R000721420009-6"

50/100j, Kell

Proteins and amino acids. XIII. Use of fluorescein-sulf酰 chloride in the determination of end amino acids of peptide chains. Bohumil Kell, Věra Kněžková, and František Šorm. *Chem. Listy* 46, 167-70 (1952). — The journal reference in *C.A.* 43, 3604c should have been *Chem. Listy* 46, 167-70 (1952) instead of *Ibid.* 167-70. R. J. C.

2A KEIL, D.

Biological Chemistry
Method - 11

Proteins and amino acids. XII. Quantitative estimation of dinitrophenylated proteins. Bohuslav Keil, Vladimír Tomášek, and Jana Nejedláková (Central Chem. Inst., Prague, Czech.). *Chem. Listy* 46, 457-61 (1952); cf. C.A. 46, 6292a. A method for the detn. of the basic end groups in proteins has been worked out on the basis of the formation of dinitrophenyl derivs. and their analysis by means of countercurrent distribution and paper chromatography. The method was applied to the detn. of 1 mol. of aspartic acid in beef serum albumin. The analysis was carried out in the dark since the dinitrophenyl derivs. of amino acids are not stable toward the daylight. M. Hudlický

KEIL, B.

Proteins. XXIX Oxidation of pancreatic proteases. p. 1837

Vol. 48, no. 12, Dec. 1954

CHEMICKÉ LISTY

Praha, Czechoslovakia

So: Eastern European Accession Vol. 5, No. 4, 1956

KEIL, B.

Proteins. XXIX. Oxidation of pancreatic proteases. In Russian. p. 471.

Vol. 20, no. 2, April 1955

SEORNIK CHEKHOVATSKIKH KHIMICHESKIKH RABOT
Praha, Czechoslovakia

So: Eastern European Accession Vol. 5, No. 4, April 1956

KEIL, B.

Inventive activity
General II

Proteins and amino acids. XIII. Activation of chymotrypsinogen to chymotrypsin. Frantisek Sorm, Bohumil Keil, and Ivan Rychlik (Central Chem. Inst., Prague, Czech). *Chem. Listy* 46, 401-4 (1952); cf. C.A. 46, 11259. On the basis of quant. paper chromatography of dinitrophenyl derivs. of amino acids, chymotrypsinogen (I) was found to contain no basic end group. Activation is followed by the formation of 2 amino groups (based on the mol. wt. 23000). In addn., a mixt. of tri- to octapeptides is formed as a result of hydrolytic processes. Cryst. α -chymotrypsin consists of 2 or 3 proteins having the same proteolytic activity. During the crystn., the content of a form constg. 1 mole of alanine and 1 mole of phenylalanine as end amino acids increases. Proteins with aspartic acid, serine, and threonine as end amino acids accumulate in the mother liquors. Activation of I is based probably on the cleavage of cyclic peptide chains which reveals the center of activity. **XIV. Enzymic activity of dinitro derivatives of α -chymotrypsin.** Frantisek Sorm and Ivan Rychlik. *Ibid.* 415-8. — By the action of 1,2,4-Cu^{II}P(NO₂)₃ 150 mg. on lyophilized chymotrypsin (I) (300 mg. in a soln. constg. 300 mg. NaHCO₃ in 15 ml. H₂O), dinitrophenyl derivs. constg. 2, 4, and 7 dinitrophenyl

groups per mol. of I, resp., were prep'd., purified by dialysis and analyzed by spec. spectrometry. Protease and chymo activities drop with increasing rate of substitution; esterase activity of I constg. 2 dinitrophenyl groups is higher, that constg. 4 groups about the same, and that constg. 7 dinitrophenyl groups per mol. of I lower than that of I. Michaelis const. and max. reaction rate of I pure and I constg. 2, 4, and 7 dinitrophenyl groups, resp., are: 0.013, 0.003, 0.002, and 0.0018; 0.50, 0.40, 0.30, and 0.12. Tendency of synthesizing polypeptides of methionine from its Pr ester increases with increasing amt. of dinitrophenyl groups. **XVI. Interaction of proteins with electrolytes.** Vladimír Kačera and Luboš Matoušek. *Ibid.* 525-8; cf. C.A. 46, 11315g. — A dynamic equil. exists in a system protein-metal ions in which protein (serum albumin) represents a complexing agent of low diffusion const. This accounts for the fact that a wave showing the reduction of ions forming a complex with serum albumin is, at a certain pH, higher than that corresponding to the flow of ions in the complex. M. Hudlický

APPROVED FOR RELEASE: 06/13/2000

Sart, J. (2)

(after crystn. from *Ca*-petr ether, m. 95°, V(5 g.) refluxed 4 hrs. with 15 ml. 40% aq. HBr, then evapd., and the residue (0.2 g.) dissolved in 10 ml. MeOH and 4 ml. *Ca*H₂N yielded 1.7 g. (55%) IV, m. 105° (from H₂O). VI (30 g.) and 22.5 g. MeCH:CHCO₂ refluxed 15 hrs. with 0.1 g. Na in 20 ml. EtOH, the mixt. acidified, extd. with three 50-ml. portions of ether, and the exts. evapd. yielded 29.5 g. (60%) of VII, b₁ 223°. The Schmidt reaction carried out in the same way as for V gave 17 g. of an oil which was directly hydrolyzed with dil. H₂SO₄ (5 ml. evapd. H₂SO₄ to 15 ml. H₂O for 15 hr.) of the oil by refluxing 5 hrs.; after the removal of the Na and SO₄ ions the residue was evapd. to dryness and dissolved in 10 ml. H₂O and 10 ml. EtOH to yield 2 g. IX, m. 151°, after crystn. from H₂O, m. 152° (decomposition). The Schmidt reaction with 30 ml. H₂SO₄, 200 ml. CHCl₃, 31 g. VIII, and 100 ml. 2.0% HNO₃ gave an instant hydrolysis, which, after hydrolysis with dil. H₂SO₄ and removal of the ions yielded 1.2 g. X, m. 155-8° (from H₂O). XIII. Use of azobenzensulfonyl chloride in determination of end amino acids of peptide chains. D. L. Yost, S. M. Vaidya, and L. G. Soren (Central Chem. In. Div., E. I. du Pont de Nemours and Co., Wilmington, Del., 1977-79, *J. Am. Chem. Soc.* 99, 1495) found it to be a suitable reagent for determining the end amino acids of peptides. The amino acid (or peptide) was reacted with 1.5-2.0 mm. and an equivalent of N₂SbCl₃ in acetone (m. 73°) at room temp., the H₂O and dioxane removed by distn., and the residue was dissolved in H₂O, dried with *CaH₂N*, then refluxed with *CaH₂N* and acidified. The amino acids found during the s. given of the following examples were: *Ala* 145°, *Glu* 145°, *Lys* 145°, *Asp* 145°, *Val* 145°, *Leu* 145°, *Asn* 212°, *Met* 212°, *Arg* 212°, *Thr* 212°, *Pro* 212°, *Trp* 212°, *Phenylalanine* 165°, *L-tryptophan* 210°, *Leu* 212°, *Leucine* 172°, *Ala* (bisulfone salt) 172°, *Ala* (bisulfone, hydrolyzed) 165°, *Ala* (aspartic acid) 218°, *Ala* (L-alanine) 171°, *Ala* (N,N-dimethyl-*Asp*) 171°, *Ala* (N,N-dimethyl-*Leu*) 171°, and *Ala* (N,N-dimethyl-*Val*) 171°. Acid hydrolysis (heating 5 hrs. at 100° with 0.1 N HCl) liberates amino acids from the peptide compound and disrupts only hydrolytic bonds so that the end amino acid could be identified from

Proteins and aminoacids. XII. Synthesis of α -methyl-glutamic acids. Jilí Šurt and František Šorm (Central Chem. Inst., Prague, Czech.), *Collections Czechoslov. Chem. Commun.*, 18, 131-9 (1953) (English summary); cf. *C.A.*, 47, 124594. — After the failure of the Curtius degradation of ester azide of (cyanomethyl)malonic acid, the Schmidt reaction was used for prep. all three α -methylglutamic acids, $\text{AcCH}_2\text{CH}_2\text{CN}$ (III) which was transferred directly to α -methylglutamic α , β (IV). IV was also obtained by hydrolysis of $\text{O}_2\text{CCMe}(\text{NH}_2)\text{CH}_2\text{CH}_2\text{CO}_2\text{Et}$ (V) resulting from the action of NH_2 on $\text{Et}_2\text{C}(\text{Ac})\text{MeCCH}_2\text{CH}_2\text{CO}_2\text{Et}$. $\text{AcCH}_2\text{CO}_2\text{Et}$ (VI) and $\text{MeCH}_2\text{CHCO}_2\text{Et}$ gave $\text{Et}_2\text{CC}(\text{Ac})\text{MeCH}_2\text{CO}_2\text{Et}$ (VII). VI and $\text{CH}_2\text{CMeCO}_2\text{Me}$ gave $\text{Et}_2\text{CC}(\text{Ac})\text{CH}_2\text{CMeCO}_2\text{Me}$ (VIII). VII and VIII were transformed to β (IX) and γ -methylglutamic (X) acids resp., by the Schmidt reaction. $\text{MeCH}_2(\text{CO}_2\text{Et})$ (17.4 g.), 16 ml. diacetone and 2 ml. Rodien's catalyst treated with 10.6 g. II at 30-40° with cooling, the mixt. acidified with HCl (1:3) after 3 hrs., treated with 25 ml. $(\text{CH}_2)_2$ and 25 ml. H_2O_2 and the aq. layer exptd. *in vacuo* to yield 95% ($\text{NCCH}_2\text{CH}_2\text{CMe}_2(\text{CO}_2\text{Et})_2$, $\text{b.p. } 110^\circ$, $\text{d}_{4}^{20} 1.0092$, $n_{D}^{20} 1.4329$, I (28.8 g.), 10 ml. dioxane, 0.5 g. Na, and 20 ml. EtOH was added, in the course of 1 hr., 21.2 g. II at 35° and the product isolated as above yielding 21 g. (61%) of III, $\text{b.p. } 118^\circ$, $\text{d}_{4}^{20} 1.0308$, $n_{D}^{20} 1.4468$. III (19.7 g.) in 120 ml. of 4.7% NH_2 soln. was added to a stirred mixt. of 30 ml. concd. H_2SO_4 in 250 ml. CHCl_3 at 20-30°, the cooled reaction mixt. dild. with 120 ml. H_2O , the aq. layer exptd. with 25 ml. CHCl_3 , refluxed 10 min., dild. with distd. H_2O to 500 ml., mixed with 216 g. $\text{Ba}(\text{OH})_2$ in 500 ml. hot H_2O , boiled 30 min., the BaSO_4 was filtered off, washed with 800 ml. hot H_2O contg. 5 ml. H_2SO_4 , the Ba and SO_4 ions removed, and the filtrate evapd. *in vacuo* to 100 ml., treated with 100 ml. EtOH and allowed to cryst. in an ice box to yield 8.8 g. (55%) of IV, m. 100° (from H_2O). To 17 g. of $\text{Et}_2\text{CC}(\text{Ac})\text{MeCH}_2\text{CH}_2\text{CO}_2\text{Et}$ in 100 ml. CHCl_3 and 34 ml. concd. H_2SO_4 was added 7 g. HgN_2 portion-wise with cooling at 30-35°, the mixt. poured onto 200 g. ice, and the CHCl_3 layer exptd. with 100 ml. H_2O_2 , dried, and evapd., leaving 14.2 g. (79%) V, m. 64-8°.

(Continued)

KEIL, B.; TOMASEK, V.; SEDLACKOVA, J.

On albumins and aminoacids. Part 14. Quantitative analysis of dinitrophenylized albumins [with summary in German]. Sbor.Chekh.khim.rab. 18 no.2:275-284 Ap '53.

(MLRA 7:6)

1. Institut organicheskoy khimii Chekhoslovatskoy Akademii nauk,
otdeleniye organicheskoy biokhimii, Praga. (Dinitrophenyl) (Albumin)

SORM, F.; KEIL, B.; RYCHLIK, I.

On albumins and aminoacids. Part 15. Activation of chymotrypsinogen to chymotrypsin [with summary in German]. Sbor.Chekh.khim.rab, 18 no.2: 285-293 Ap '53. (MLRA 7:6)

1. Institut organicheskoy khimii Chekhoslovatskoy Akademii nauk, otdeleniye organicheskoy biokhimii, Praga. (Chymotrypsin)

KEIL, B.; SORM, F.

On proteins. Part 19. Methylation of chymotrypsogen and chymotrypsin
[with summary in English]. Sbor.Chekh.khim.rab. 18 no.4:550-559 Ag '53.
(MLRA 7:6)

1. Department of Organic Biochemistry, Central Chemical Research Institute,
Prague. (Chymotrypsin) (Methylation)

KEIL P.

Proteins. XIX. Methylation of chymotrypsinogen and chymotrypsin. p.245
(Chemicke Listy. Vol. 47, No 2, Feb. 1953) Czéchoslovakia

SO: Monthly List of East European Accessions, Vol. 2, #8, Library of Congress,
August 1953, Incl.

Keil, B.

✓ Herout, V., Keil, B., Protiva, M., Hudlicky, M., Ernest, I., and Gut, J.: Laboratorní technika organické chemie. Prague: Nakl. CSAV, 1954. 756 pp. Kčs 80. Reviewed in Chem. Listy 49, 1415(1955).

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Herout, V., Keil, B., Protiva, M., Hudlicky, M., Ernest, I., and Gut, J.: Organic Chemistry Laboratory Techniques. Publishing House CSAV. 1954. 756pp. Kčs. 86. Reviewed in Chem. Listy 49, 1415(1955).

PM/ek

LAUCIKOVA, O.; JAKUBOVIC, A.; KAMIL, B.; SORM, F.

Viruses I; isolation and chemical properties of Rous sarcoma.
Chekh. biol. 3 no.5:298-307 Nov 54.

1. Institut organicheskoy khimii ChSAN, organicheskaya biokhimiya,
Praga.

(VIRUSES,
Rous sarcoma virus, isolation & chem.)
(NEOPLASMS, experimental,
Rous sarcoma virus, isolation & chem.)
(SARCOMA, experimental,
Rous sarcoma virus, isolation & chem.)

KEIL, F.

MELOUN, B.; KEIL, B.; SORM, F.

Amino acids and peptides. Part 9. Constitution of the peptide phalloidine; part 2 [in German with summary in Russian]. Sbor.Chek.khim.rab. 19 no.1:153-161 F '54. (MLRA 7:6)

1. Otdeleniye organiceskoy biokhimii, Institut organiceskoy khimii Chechoslovatskoy Akademii nauk, Praga. (Phalloidine)

Author: HORVAT, F.; KNESSLOVA, V.

Czechoslovak. Chem. Comm. (Collection of Czechoslovak Chemical Communication. Praha. Vol. 19, no. 4, Aug. 1954)
(Collection of Czechoslovak Chemical Communication. Praha. Vol. 19, no. 4, Aug. 1954)
Czechoslovak Inst. of Electronic Association (Cesos), Brno, Czech. Rep.,
Czechoslovakia.

KETL, B.

2

4956* Proteins. O belkakh. XXV. Chemical Structure of Proteins. Introduction. O khimicheskom stroye belkov. Vvedenie. F. Shorn. XXVI. Photometric Analysis of Protein Hydrolysates. Fotometricheskii analiz belkovikh gidrolizatov. (Russian.) B. Kell. Collection of Czechoslovak Chemical Communications, Volume 5, Oct. 1954, p. 1003-1017.
Analysis of individual peptides or peptide fractions as an approach to the problem. Tables, graphs. 29 ref

11/11/71

11/11/71

Communication LXVI. Radioactive analysis of organic compounds. 1. 6. (Collection of
Czechoslovak Chemical Communication, Praha, Vol. 19, no. 5, Oct. 1954)
by V. Vojtěchovský, J. Černý, A. Štěpnička, (L. L., I., J. L. Černý),
T. Štěpničková, J. Štěpnička.

KIWI, E.

FORM : FORM, F.

On rotation. XXVII. Comparative study of the stability of the 1,2-dihydro-
2H-pyridine-1,4-dioxane (Dypridoxine). p. 1-10 (Collection of
Czechoslovak East Chemical Communication. Praha. Vol. 19, no. 5, Oct. 1954)
IC: 1954-1955/Chemical Abstracts (1955), 47, 11, 111.
Type "F", "cl.

1967, KRL
XEL, Remy

Proteins XXVI. Photoactive enzymes of plants
dehydrates bovine hemoglobin 100%
Chem. News 64 11/5/64 p 45 177-181
quartz cells
tear by hand
negative
directed light
state is left
negative
enzymes
enzymes
hydrogen, and feed green plants
M. Hudlicky
examples.

KEIL, B.

"Proteins. XXVII. Comparative study of acidic peptide fractions from partial hydrolyzates of chymotrypsin and trypsin." Ceskoslovenska Morfologie, Praha, Vol. 48, No. 5, May 1954, p. 735.

SO: Eastern European Accessions List, Vol. 3, No. 11, Nov. 1954, I.C.

KEIL, Bořivoj

CZECH

Proteins. XXVIII. Comparison of the arginine peptides from partial hydrolyzates of chymotrypsinogen and trypsinogen. Jarl Vaněk, Bořivoj Keil, and František

Bořivoj Keil, akad. věd. Pražské České vědecké Akademie, 46, 1977, 16(1967), 111-117. — Chymotrypsinogen (I), trypsinogen (II), and insulin (III) (in 200 cc. saline) were partially hydrolyzed by heating 6 days at 37° with 10 ml. concd. HCl. The arginin peptides were sepd. by passing the hydrolyzates over a column filled with Amberlite IRA-400, and further investigated by means of total hydrolysis and the dinitrophenyl method. Both I and II contain the grouping asyl-arginyl and valyl-arginyl. I differs from II by contg. a third mol. of arginine bound to threonine and alanine. In III the main grouping of glycine and arginine corresponds to the Saenger formula but the grouping of arginine with leucine is in contradiction.

XXIX. Oxidation of pancreatic proteases. Bořivoj Keil. *Ibid.* 1837-41. — Oxidation of I, II, and α - and γ -chymotrypsins with performic acid is accompanied by denaturation of the proteins and by splitting off small amts. of low-mol. peptides. Hydrolytic fission of peptide bonds during the reaction cannot be prevented. Biol. activity decreases rapidly with oxidation. Two main chains of chymotrypsin seem to be linked not only by disulfide bridges, but by stronger, probably peptide bonds. The oxidation was carried out by dissolving the protein (1 g.) in a mixt. of 40 ml. 90% HCO₂H and 4 ml. 26% H₂O₂, holding at room temp. 20 min., adding 40 ml. H₂O, and evapg. to dryness below 50° *in vacuo*. The residue was stirred with Me₂CO (60 ml.), the gummy mass centrifuged, washed twice with 60 ml. Me₂CO, stirred with 60 ml. 0.1*N* NH₄OH, after 2 hrs. the pH adjusted to 8 with 5*N* H₂SO₄, the ppt. 0.8 g. centrifuged, the filtrate acidified with 6*N* AcOH to pH 4, dried

S. Vanecek

from frozen state, the residue dissolved in 8 ml. H₂O, treated with 8 ml. 50% AcONH₄, the ppt. of 20 mg. centrifuged, the soln. evapd., and AcONH₄ evapd. at 90° and 0.1 mm., leaving 80 mg. product (A), sol. in H₂O, mol. wt. approx. 3000. Ultraviolet absorption spectra show an absence of aromatic amino acids. Ultraviolet spectra of I, II, and fractions (A) of both proteins are given.

M. Hudlicky

2/2

KETE, B

Proteins, XXIV. Substitution of ϵ -amino groups of
lysine in the molecule of chymotrypsinogen by the reaction
with 2,4-dinitrofluorobenzene. Vlada Kuncalová, Bohumil Kral
and František Šorm (Ústav Org. Chem., České Vysokého
Czech.), Česk. Lidsk. 48, 500-501 (1959); cf. C. A. 52,
64788.—By carrying out the reaction of 2,4-dinitrofluorobenzene (I) with chymotrypsinogen in a soln of NaHCO_3 and 1% Me_2N , under certain conditions all of the ϵ -amino groups in lysine reacted with I. The importance of pH in the reaction is stressed.
M. Hudlicky

13
CZECH

The peptide accompanying pancreatic ribonuclease acid
K. Kopecký and M. Hrubáková (Česk. Akad. věd, Prague,
Časop. Lékař 49, 271-5 (1955)) — The peptide accompanying
pancreatic ribonuclease acid was found to give a slight
acidic reaction. — M. Hrubáková

✓ Proteins. XXXIII. Differences in the arginine peptides of some serum albumins. Věra Kněsllová, Vladimír Kostka, Bohumír Keil, and František Sorm (Czech. Akad. Věd, Prague). *Chem. Listy* 49, 913-20 (1955); cf. *C.A.* 49, 10009b. —Human, beef, horse, duck, and sheep serum albumins were subjected to partial hydrolysis by heating 200-mg. portions of the proteins 124 hrs. at 37° with 10 ml. concd. HCl. From the partial hydrolyzates the arginine peptides were isolated by means of the ion exchanger Amberlite IR-A. Hydrolysis of the arginine peptides with equal vols. of concd. HCl (16 hrs. at 105° in a sealed tube), paper chromatography in BuOH-AcOH system, hydrolysis of individual fractions with 6*N*-HCl (16 hrs. at 105°), paper chromatography in the BuOH-AcOH system, and dinitrophenyl analysis revealed considerable differences in the content of the individual amino acids in serum albumins of various origins, although the total hydrolyzates of all of the investigated serum albumins showed only slight differences.

M. Hudlický

KEIL, BORIVOT

Virus studies. VII. A comparative study of two strains of the tobacco mosaic virus. Alexander Jakubovič, Libor Šlechta, Bohumil Keil, and František Šorm (Czech. Akad. Věd, Prague). *Chem. Listy* 49, 1501-4 (1955); cf. *C. d.* 49, 10410s. —Ultraviolet spectrum, electrophoretic mobility, and qual. and quant. content of amino acids are given for cryst. forms of the ordinary strain and Alke-strain of the tobacco mosaic virus. Only the Alke-strain was found to contain histidine. Electron-microscope photographs of both cryst. forms are given. M. Hudlický

(4)

Keil, DORIVOJ

Met ✓ Virus studies. VII. A comparative study of two strains
of the tobacco mosaic virus. Alexander Jakubovit, Libor
Slechta, Bohumil Keil, and František Sorm. Collection
Czechoslov. Chem. Commun. 21, 29-32 (1950) (in German).—
See C.A. 50, 440c. *4*
E. I. C.

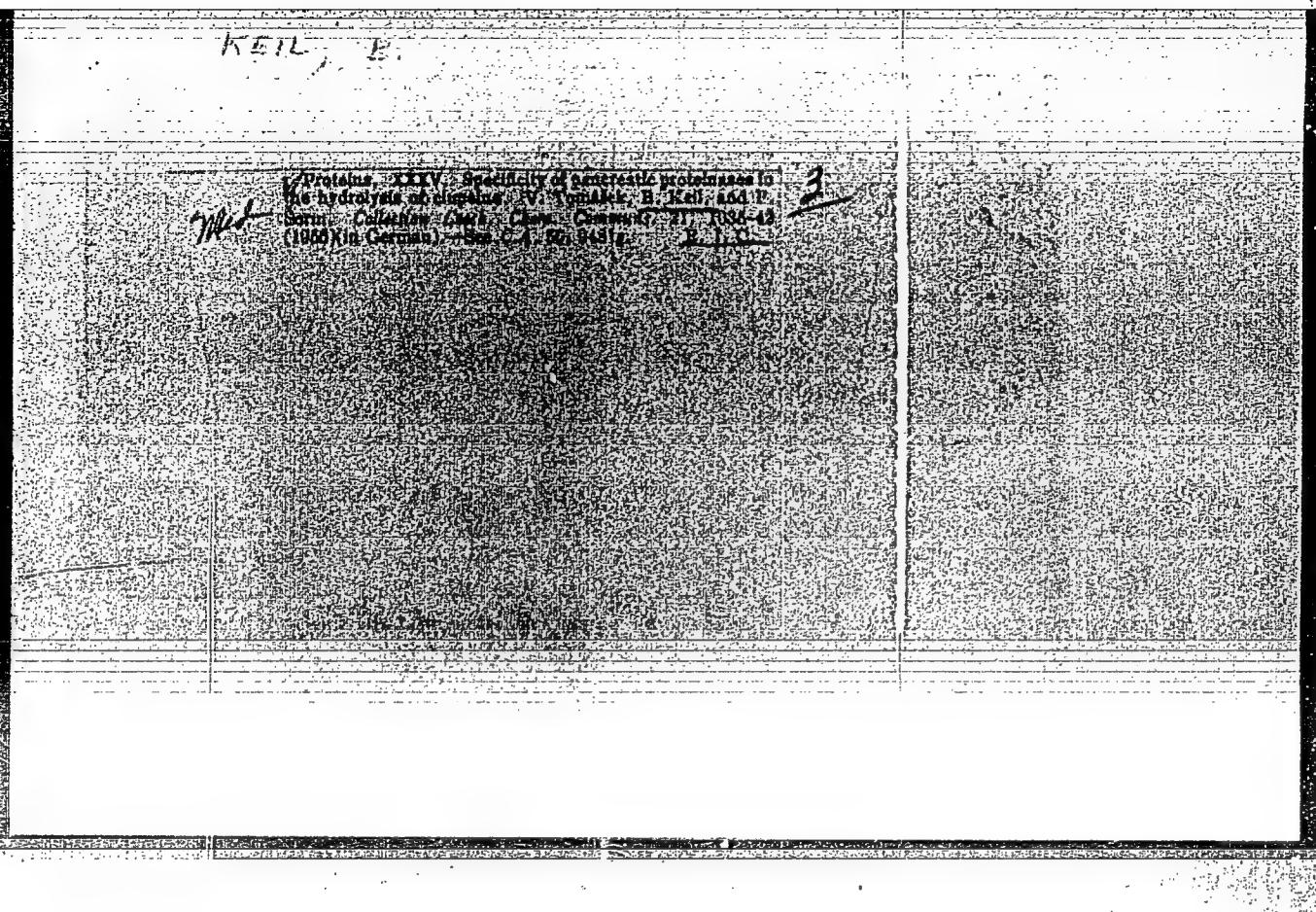
...MIL, B.

MIL, B. Proteins. XXXV. Specificity of pancreatic proteinases in
the fission of clupein. p. 623. Vol 50, no. 4, Apr. 1956.
CENZKE LISTY. Praha, Czechoslovakia.

SOURCE: East European Accessions List (EEAL) Vol. 6, No. 4—April 1957

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000721420009-6



APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000721420009-6"

KEYL', B.

International conference on protein. Vop. med. khim. 3 no.1:74-77
Ja-F '57 (MLRA 10:4)

1. Khimicheskiy institut Chechoslovatskoy akademii nauk, Praga.
(PROTEINS)

KEL, B.

AUTHORS: Štrna, P.; Neill, B.; Holešovský, V.; Nejoun, B., Mikš, O., and Šindelk, J.

CZ/8/52(82)/10-26/59

TITLE: Proteins. XIII. Comparison of the Microstructure of Chymotrypsinogen and Trypsinogen. Preliminary Communication (O. Miklovič, K. Štrna, V. Štrna and J. Šindelk, submitted)

PUBLISHER: Praha, 1956, Vol. 92(82), No. 10, pp. 1992-1995

(Czechoslovakia)

ABSTRACT: This is a continuation of the discussion on the micro-structure of proteins in which the authors drew on their own experimental results (previously published) and those of others. Attention is drawn to the repetition of certain peptide residues in the two proteins considered. There are 5 tables and 34 references, 12 of which are Czech, 22 German.

CARD 1/2

6

ASSOCIATION: Biologické oddělení, Českého listav, Československá akademie věd, Praha, Biochemistry Division, Institute of Chemistry, Czechoslovak Academy of Science, Prague

SUBMITTED: March 13, 1956

CARD 2/2

KELL, B.; SUKES, F.; MARINK, F.

Proteins and amino acids. XXXVIII. Comparison of arginine peptides of partial hydrolysates of two hemoglobins. p. 352. (Chemicke Listy, Vol. 51, no. 2, Feb. 1957.)

SO: Monthly List of East European Accession (EEAL) Vol. 6, no. 7, July 1957. Uncl.

KEIL, B.

Abs Jour: Ref Zhur.-Khimiya, 1958, No II

Author: O. Mikes, J. Vanecek, B. Meloun, B. Keil, V. Kostka, J. Kara.

Inst: Not given

Title: Multiple-Chamber Appliance for the Preparative Electro-phoresis.

Orig Pub: Chem. listy, 1957, 51, No 8, 1562-1569.

Abstract: A description of a modified multi-chamber appliance ~~xi~~ for the preparative zonal electrophoresis at the constant value of pH, in which are combined ~~x~~ the advantages of 3-chamber Svenson's appliance with those electrophoretical ones to the work in an auxiliary medium. A rectifier with a regulated voltage of 0-10,000 v serves as a source of tension.

KEIL, Borivoj

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000721420009-6"

CZECHOSLOVAKI: / Laboratory Equipment, Instruments, Theory, Construction, Use.

Abs Jour : Ref Zhur - Khim., No 15, 1958, No 50138

Author : Kocont, Alexandr; Brada, Zbynok; Keil, Borivoj.

Inst : Not given

Title : Gravimetric Fraction Collector for Chromatography.

Orig Pub : Chem. listy, 1957, 51, No. 8, 1575-1576.

Abstract : An electromagnetic arrangement, which enhances the reliability of the work of a gravimetric lever collector of fractions, is described. -- M. Ryba.

HEIL, B.; SUDL, F.

"Proteins." XLVII. Cysteic-acid peptides from a partial chymotrypsinogen hydrolysate. XLVIII. Cysteic-acid peptides from a partial trypsin hydrolysate. In English. p. 1558.

C. LISTING OF CZECHOSLOVAK CHEMICAL COMMUNICATIONS, Praha, Czech., Vol. 24, No. 5, May 1959

Monthly List of East European Accessions (EDAL), LC, Vol. 8, No. 6, Sept. 59

Unclassified

MELOUN, B.; HOLEYSOVSKY, V.; VANECK, J.; KEIL, B.; SORM, F.

Proteins. LIII. Peptides of aspartic acid and glutamic acid isolated from a chymotrypsinogen hydrolysate. In English. Coll.Cz.Chem. 24 no.9:3002-3006 S '59. (EEAI 9:5)

1. Department of Biochemistry, Chemical Institute, Czechoslovak Academy of Science, Prague.

(Proteins) (Peptides) (Aspartic acid) (Glutamic acid)
(Chymotrypsinogen)

VANECEK, J.; KEIL, B.; MELOUN, B.; SORM, F.

Proteins LIV. Isolation of some peptides from tryptic hydrolysates
of chymotrypsinogen and diisopropylphosphoryltrypsin. In English.
Coll.Cz.Chem. 24 no.9:3148-3153 S '59. (EEAI 9:5)

1. Department of Biochemistry, Czechoslovak Academy of Science,
Prague.

(PROTEINS) (PEPTIDES) (CHYMOTRYPSINOGEN)
(TRYPSIN DIISOPROPYL PHOSPHATE)

KARADZOVÁ, M.; KEIL, B.; SORM, F.

Peptides isolated from acid hydrolysate of edestin. Coll Cz Chem 25
no.11:2878-2888 N '60. (EEAI 10:6)

1. Institute of Organic Chemistry and Biochemistry, Czechoslovak
Academy of Science, Prague.
(Peptides) (Edestin)

PRUSIK, Z.; KEIL, B.

An investigation of conditions of separating substances by high-voltage electrophoresis on paper. Coll Cz Chem 25 no.8:2049-2058
(EEAI 10:9)
Ag '60.

1. Department of Biochemistry, Institute of Chemistry, Czechoslovak
Academy of Science, Prague.

(Electrophoresis) (Separation)

VANECEK, J.; MELOUN, B.; KOSTKA, V.; KEIL, B.; SORM, F.

Proteins. LXI. Peptides isolated from peptic hydrolysate of chymotrypsinogen. Coll Cz Chem 25 no.9:2358-2368 S '60.
(EEAI 10:9)

1. Institute of Organic Chemistry and Biochemistry Czechoslovak Academy of Science, Prague.

(Proteins) (Peptides) (Chymotrypsinogen)

SORM, F.; KEIL, B.; VANECEK, J.; TOMASEK, V.; MIKES, O.; MELOUN, B.;
KOSTKA, V.; HOLEYSOVSKY, V.

Proteins. LXIII. Lower structures in the chains of proteins. Coll Cz
chem 26 no.2:531-578 F '61. (EEAI 10:9)

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Academy of Science, Prague.

(Proteins)

HEYROVSKY, Jaroslav, dr., akademik, nositel Nobelovy ceny; JANAK, Jaroslav, inz.; VOLF, Milos Bohuslav, dr.; KELL, Borivoj, Dr.Sc., laureat statni ceny; KOSSLER, Ivo, dr.

Observations of our famous collaborators on making new laboratory instruments. Tech praca 14 no.8:655-664 Ag '62.

1. Ceskoslovenska akademie ved (for Janak and Kossler).

S/058/62/000/012/033/048
A160/A101

AUTHORS: Keil, B., Šorm, F.

TITLE: On proteins. LXXI. An analysis of protein structures from the aspect of amino acid interchanges

PERIODICAL: Referativnyy zhurnal, Fizika, no. 12, 1962, 36 - 37, abstract 12D260 ("Collect. Czechosl. Chem. Commun.", no. 5, 1962, v.27, 1310 - 1319, English; summary in Russian)

TEXT: The series of amino acids is compared to various proteins to find the symmetry or a similarity in the arrangement of amino acids along the polypeptide chain by taking into consideration a possible substitution of one amino acid by another. By using the method of symbols, it was shown that the symmetry principle occurs rather frequently in the arrangement of amino acids. For ribonuclease and Cytochrome it was discovered that a definite series type is most frequently found, i.e., a major part of the protein is built from several main series. This is particularly clearly apparent in the case of α - and β -chains of hemoglobin. Part 70 see ref. 12D259.

[Abstracter's note: Complete translation]

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Inst. Organic Chem. & Biochem., Czech AS, Prague

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On proteins. Part 73: Desulfuration of sulfur containing
amino acids in peptides. Coll Cz Chem 27 no.7:1673-1677
Jl '62.

1. Institute of Organic Chemistry and Biochemistry, Czechoslovak
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On proteins. Part 74: Hydrogenation of aromatic amino acids
in peptides. Coll Cz Chem 27 no.7:1678-1686 Jl '62.

1. Institute of Organic Chemistry and Biochemistry, Czechoslovak
Academy of Sciences, Prague (for Keil and Sorm). 2. Institute
of Physical Chemistry, Czechoslovak Academy of Sciences, Prague
(for Zikan). 3. Institute of Chemistry, Slovak Academy of
Sciences, Bratislava (for Rexova).

KEILL, B.; MORAVEK, J.; DLOUHA, V.; FILIP, J.

On proteins. Part 75: Desulfuration and hydrogenation of
amino acids by using tritium. Coll Cz Chem 27 no.7:1687-1691
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1. Institute of Organic Chemistry and Biochemistry, Czechoslovak
Academy of Sciences and Institute for Research, Production and
Utilization of Radioisotopes, Prague.

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KEILOVÁ, H; KEIL, B.

Czechoslovakia

Institute of Organic Chemistry and Biochemistry,
Czechoslovak Academy of Sciences -- Prague
- (for all)

Czechoslovak
Prague, Collection of Chemical Communications,
No 9, 1962, pp 2186-2191

"Proteinases of the Ehrlich Ascites Tumour. I.
Basic Characterization of the Proteinases and
Their Dependence on the Growth of the Tumour."

KEILOVÁ, H; KEIL, B.

Czechoslovakia

Institute of Organic Chemistry and Biochemistry,
Czechoslovak Academy of Sciences -- Prague
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Communications, No 9, 1962, pp 2193-2200

"Proteinases of the Ehrlich Ascites Tumour. II
Separation of Proteinases of the Ascites Fluid
and Ascites Cells."

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Proteinases of the Ehrlich ascites tumor. Part 1: Basic characterization of the proteinases and their dependance on the growth of the tumor. Coll Cs Chem 27 no.9:2186-2192 S '62.

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1. Institute of Organic Chemistry and Biochemistry, Czechoslovak
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1. Institute of Organic Chemistry and Biochemistry, Czechoslovak Academy of Sciences, Prague.

KEIL, B

CZECHOSLOVAKIA

MELOUN, B; KOSTKA, V; KEIL, B; SORK, F.

Institute of Organic Chemistry and Biochemistry of the
Czechoslovak Academy of Sciences, (Prague (for all)

Prague, Collection of Czechoslovak Chemical Communications,
No 10, 1963, pp 1749-2777

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Sulpho-Chymotrypsinogen Insoluble in Acid Environment."

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1. Institute of Organic Chemistry and Biochemistry of the Czechoslovak Academy of Sciences, Prague. 2. Permanent address: Institute of Natural Substances of the Academy of Sciences of the U.S.S.R., Moscow (for Jegorov). 3. Advisory Board Chairman, "Collection of Czechoslovak Chemical Communications" (for Sorm). Submitted February 5, 1964.

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KEIL, B.

MILKOVA, L; KOPEC, Z; KEIL, B

1. Institute of Chemistry, Slovak Academy of Sciences,
Bratislava - (for ?); 2. Institute of Organic Chemistry
and Biochemistry, Czechoslovak Academy of Sciences,
Prague - (for ?)

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KEV/EL

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